EXERGY ANALYSIS OF GAS-TURBINE COMBINED CYCLE WITH CO₂ CAPTURE USING AUTO-THERMAL REFORMING OF NATURAL GAS

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CO₂ capture

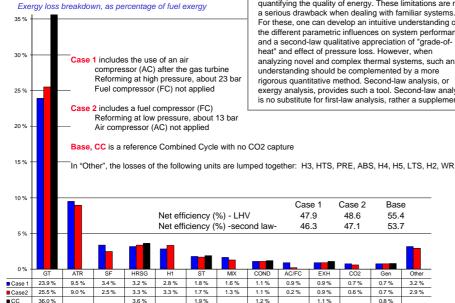
A concept for capturing and sequestering CO2 from a natural-gas fired combined-cycle power plant is presented.

The present approach is to de-carbonise the fuel prior to combustion by reforming natural gas, producing a hydrogen-rich fuel. The reforming process consists of an air-blown pressurised auto-thermal reformer (ATR) that produces a gas containing H₂, CO and a small fraction of CH4 as the combustible components. The gas is then led through a water-shift reactor (HTS, LTS), where the equilibrium of CO and H2O is shifted towards CO2 and H2. The CO₂ is then captured from the resulting gas by chemical absorption (ABS). The gas turbine of this system is then fed with a fuel gas containing approximately 50% $\rm H_2.$ A very important aspect of this type of process is the integration between the combined cycle and the reforming process

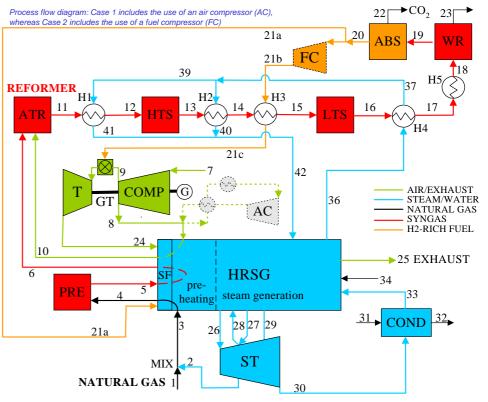
A model of the gas turbine GE9351FA from General Electric was used in the simulations.

Computational results. Cases explained in diagram below

	Case1	Case2	Base
Natural gas LHV (MW) (1)	879	864	683
Air extracted to ATR (kg/s) (8)	83.2	79.5	
ATR outlet pressure (bar) (11)	22.8	12.8	
ATR inlet (kg/s) (6+10)	141.3	117	
Fuel composition (%) (21c)			
H ₂	55.6	56.3	
N ₂ +Ar	41.2	40.7	2.0
со	0.3	0.4	
CO ₂	2.0	2.0	1.0
CH ₄	0.5	0.4	93.0
C ₂ H ₆			4.0
H ₂ O	0.3	0.2	
Fuel flow to GT (kg/s) (21c)	67.9	67.4	14.6
Fuel flow to SF (kg/s) (21a)	7.8	5.6	
Power output GT (MW)	253.	256	243.
Power output ST (MW)	179	181	140
Auxiliary power (MW)	5.1	5.1	4.6
Air / fuel compression (MW)	5.5	10.8	
Net power output (MW)	422	421	378
Net efficiency (%) – LHV	47.9	48.6	55.4
Net efficiency (%) -second law-	46.3	47.1	53.7
CO ₂ emissions (g _{CO2} /kWh _{el})	57	56	365
CO ₂ reduc. vs. Base (%/kWh _{el})	84.4	84.7	



Gas-turbine Combined Cycle with Auto-thermal reforming



TURBINE, COMP = COMPRESSOR, AC = AIR COMPRESSOR - SUPPLEMENTARY FIRING HRSG - HEAT RECOVERY STI ATR = AUTO-THERMAL REFORMER, HTS = HIGH TEMPERATURE SHIFT-REACTOR, WR = WATER REMOVAL, ABS = CO2 ABSORBER, FC = FUEL COMPRESSOR

Exergy analysis

Traditional first-law analysis, based upon unit-performance characteristics coupled with energy balances, invariably leads to a correct final answer. However, such an analysis cannot locate and quantify the losses that lead to the obtained result. This is because the first law embodies no distinction between work and heat, no provision for quantifying the quality of energy. These limitations are not a serious drawback when dealing with familiar systems. For these, one can develop an intuitive understanding of the different parametric influences on system performance and a second-law qualitative appreciation of "grade-ofheat" and effect of pressure loss. However, when analyzing novel and complex thermal systems, such an understanding should be complemented by a more rigorous quantitative method. Second-law analysis, or exergy analysis, provides such a tool. Second-law analysis is no substitute for first-law analysis, rather a supplement

Case 2

48.6

47.1

0.7 %

0.6 %

Rase

55.4

53.7

0.7 %

0.8 %

3.2 %

2.9 %

Exergy analysis

Exergy balance, including the loss (Irreversibility) T_0 $\dot{m}_k e_k + \dot{W} + \dot{I}$ m; e; O_1 1 $\overline{T_l}$ Flow exergy Heat exchange Flow exergy Work out of system Ir Irreversibility into system

Computational tools: GTPRO 10.0 (Thermoflow, Inc.) and PRO/II 5.11 (Simsci, Inc.).

Conclusions

Two cases of a gas-turbine combined-cycle power plant with natural-gas reforming, $\rm CO_2$ capture, and combustion of a hydrogen-rich fuel were simulated. The resulting **first**law (LHV) efficiencies were 47.9 and 48.6, respectively. the second-law (exergy) efficiencies were 46.3 and 47.1. A comparable conventional natural-gas fired combined cycle gave first- and second-law efficiencies of 55.4 and 53.7. respectively.

If was seen that a lower pressure (approx. 14 bar) in the reforming process and fuel compression was beneficial from a thermodynamic point of view compared to maintaining a high pressure (approx. 25 bar) throughout the process

The irreversibility (exergy loss) was determined in each unit of the system, see diagram to the left. The greater loss in the new concepts resulted from additional losses in supplementary firing (SF) and heat exchange (H1-H3) tween the reforming and power cycle process

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